# Isolation and Partial Synthesis of $3(R)$ - and $3(S)$-Deoxypumiloside; Structural Revision of the Key Metabolite from the Camptothecin Producing Plant, Ophiorrhiza pumila 

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#### Abstract

Both the C-3 epimeric pair of $3(R)$ - and $3(S)$-deoxypumiloside were found in Ophiorrhiza pumila (Rubiaceae), a source plant of camptothecinoid metabolites. These structures were confirmed by spectroscopic analysis and partial stereoselective syntheses. The configuration at $\mathrm{C}-3$ of the previously reported "deoxypumiloside" is revised to $3(R)$ from 3(S). © 1997 Elsevier Science Ltd.


Camptothecin (1) is a natural molecule well-known for its potent biological properties such as inhibitory activities against tumor cells and DNA topoisomerase $I^{1}$ and activity against HIV-1.2 From a biogenetic point of view, camptothecin (1) possessing a quinoline skeleton has been shown to be formed from the indole alkaloid, strictosidine (2). ${ }^{3-5}$ Although strictosamide (3), a lactam derivative of strictosidine, was reported to be a biogenetic precursor of camptothecin, ${ }^{3-5}$ "poststrictosamide biosynthetic events", so named by Hatchinson, have not yet been clarified. During our chemical investigation of camptothecin (1), we have found


Camptothecin (1)


Pumiloside (4)

$3 \mathrm{H}-\beta: 3(R)$-Deoxypumiloside (5)
("Deoxypumiloside")
$3 \mathrm{H}-\alpha: 3(S)$-Deoxypumiloside (6)

that Ophiorrhiza pumila (Rubiaceae) produces not only camptothecin but also a variety of camptothecinrelated alkaloids. ${ }^{6-8}$ Among them, pumiloside (4) ${ }^{7}$ and deoxypumiloside ${ }^{7}$ were considered to be biogenetic intermediates to camptothecin (1). Pumiloside (4) was also found in Camptotheca acuminata by Hecht and reported independently. ${ }^{9}$ The presence of these hybrid-type molecules suggested that camptothecin formation from strictosamide (3) starts from the $A / B$ ring conversion of the indole to a quinoline skeleton followed by $D$ and E ring transformations. In order to find new secondary methabolites, which would produce further evidence for the camptothecin biosynthesis, an exhaustive investigation of the constituents in $O$. pumila was carried out. In this paper, we describe the isolation of both $3(R)$ - and $3(S)$-deoxypumiloside as their tetraacetates and the unequivocal structural clarification by spectroscopic and synthetic methods.

Acetylation of a crude natural deoxypumiloside fraction, which was isolated from $O$. pumila, led to the isolation of two acetates ( $7^{10}$ and $8^{11}$ ) in the ratio of $3: 1$. Both of the products exhibited the same UV absorptions ( $320,313,306,300,293,236,204 \mathrm{~nm}$ ) and the molecular formula ( $\mathrm{C}_{34} \mathrm{H}_{36} \mathrm{~N}_{2} \mathrm{O}_{12}$ ) which agreed with the anticipated data of deoxypumiloside tetraacetate. The ${ }^{13} \mathrm{C}$-NMR spectra of 7 and 8 were very similar except for the chemical shifts of the C-15 carbons (7: $\delta 28.3 \mathrm{ppm}, \mathbf{8}: \delta 23.8 \mathrm{ppm}$ ). Furthermore, the CD spectra of 7 and 8 showed the opposite cotton effect in the region between $320-270 \mathrm{~nm}$. These data clearly show 7 and 8 as epimeric isomers; most likely they are epimers at C-3. For elucidation of their configurations, careful NOE experiments were done. An irradiation of $15-\mathrm{H}(\delta 3.08)$ in 7 led to enhancement ( $10 \%$ ) of the peak intensity of $3-\mathrm{H}(\delta 5.01)$, indicating that this compound has a $3(R)$-configuration ( $3 \mathrm{H}-\boldsymbol{\beta}$ ). On the other hand, an irradiation of $3-\mathrm{H}(\delta 4.73)$ of the minor compound 8 led to enhancement $(4 \%)$ of the peak intensity of $19-\mathrm{H}(\delta$ 5.80 ), indicating that it is the $3(S)$-congener $(3 \mathrm{H}-\alpha)$.


To confirm these structures, we next undertook the partial syntheses of both deoxypumilosides using vincoside lactam (9) and strictosamide (3) as the starting materials, respectively. 3(R)-Pumiloside tetraacetate (10) was prepared from 9 by a three-step operation in $65 \%$ yield. ${ }^{7}$ Thus, 10 was treated with LDA and then with $N$-phenyltrifluoromethanesulfonimide ${ }^{12}$ in THF-HMPA at $-78 \sim 0^{\circ} \mathrm{C}$ to give the enol triflate 11 in $89 \%$ yield. In the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum, the peak of $N_{\mathrm{a}}-\mathrm{H}$ disappeared and the methylene protons on C-5 shifted to a lower field ( $\delta 5.50,4.82$ ) compared with those of $10(\delta 5.08,4.55)$. UV absorptions of $11(321,308,236,205$ nm ) were similar to those of $3(R)$-deoxypumiloside tetraacetate (7). These observations indicated that the trifluoromethanesulfonyl group was introduced to the oxygen at the C-7 position to form enol triflate. 11 thus

obtained was treated with palladium acetate, 1,1'-bis(diphenylphosphino)-ferrocene (DPPF), triethylamine and formic acid ${ }^{13}$ in dioxane at $60^{\circ} \mathrm{C}$ to afford the deoxygenated compound 7 in $80 \%$ yield. In the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum, a singlet peak due to $7-\mathrm{H}$ was observed at $\delta 8.07$. Spectroscopic data (UV, ${ }^{1} \mathrm{H}-,{ }^{13} \mathrm{C}-\mathrm{NMR}, \mathrm{MS}, \mathrm{CD}$ ) of the synthetic compound were identical with those of the acetate of deoxypumiloside that we obtained from O. pumila and was reported in Tetrahedron Letters in 1990. ${ }^{7}$ The present results clearly indicate that the formerly deduced stereochemistry is erroneous and the configuration of $\mathrm{C}-3$ of "deoxypumiloside" should be revised from $\mathrm{C}-3(S)$ to $\mathrm{C}-3(R)$. This conclusion is further substantiated by a parallel stereoselective conversion. $3(S)$-Deoxypumiloside tetraacetate (8) was prepared from strictosamide (3), which possesses the $3 \alpha-\mathrm{H}$ configuration, via the reductive deoxygenation at the $\mathrm{C}-7$ position in $3(S)$-pumiloside tetraacetate (12). The synthetic compound 8 was identified as $3(S)$-deoxypumiloside tetraacetate, the minor acetate that was obtained as the minor congener during acetylation of the crude "deoxypumiloside" fraction of $O$. pumila. From these synthetic studies, the absolute stereochemistry of both deoxypumilosides was unambiguously established.

In conclusion, we found that $O$. pumila produces both the $3(R)$ - and $3(S)$-deoxypumiloside ( 5 and 6 ). We now abandon the name "deoxypumiloside", and this name appearing in previous literature ${ }^{6,7}$ should be changed to read $3(R)$-deoxypumiloside (5) hereafter. The structure of each tetraacetate compound including the absolute configuration was confirmed by spectroscopic data and partial syntheses. The findings that both the $3(R)$ - and $3(S)$-deoxypumiloside are present in $O$. pumila and that $3(R)$-deoxypumiloside is more richly abundant than the $3(S)$ congener are quite important in further clarification of the camptothecin biosynthesis.

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10. 3(R)-Deoxypumiloside tetraacetate (7): HR-MS (NBA) m/z: 665.2357 (Calcd for $\mathrm{C}_{34} \mathrm{H}_{37} \mathrm{~N}_{2} \mathrm{O}_{12}: 665.2347$ ); FABMS (NBA) $m / z(\%): 665\left(\mathrm{MH}^{+}, 60\right), 154(100) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 8.08(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}, 12-\mathrm{H}), 8.07$ $(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.83(\mathrm{dd}, 1 \mathrm{H}, J=8.0,1.2 \mathrm{~Hz}, 9-\mathrm{H}), 7.72(\mathrm{ddd}, 1 \mathrm{H}, J=8.6,7.1,1.5 \mathrm{~Hz}, 11-\mathrm{H}), 7.55(\mathrm{ddd}, 1 \mathrm{H}, J=8.2$, $7.1,1.3 \mathrm{~Hz}, 10-\mathrm{H}), 7.53(\mathrm{~d}, 1 \mathrm{H}, J=2.7 \mathrm{~Hz}, 17-\mathrm{H}), 5.47(\mathrm{ddd}, 1 \mathrm{H}, J=17.3,9.9,9.9 \mathrm{~Hz}, 19-\mathrm{H}), 5.34(\mathrm{~d}, \mathrm{IH}, J=1.9$ $\mathrm{Hz}, 21-\mathrm{H}), 5.33-5.29(\mathrm{br}-\mathrm{d}, 1 \mathrm{H}, J=17.1 \mathrm{~Hz}, 5-\mathrm{H}), 5.31(\mathrm{dd}, 1 \mathrm{H}, J=17.1,2.0 \mathrm{~Hz}, 18 \mathrm{~B}-\mathrm{H}), 5.27(\mathrm{dd}, 1 \mathrm{H}, J=9.4,9.4$ $\left.\mathrm{Hz}, 3^{\prime}-\mathrm{H}\right), 5.18(\mathrm{dd}, 1 \mathrm{H}, J=10.0,1.9 \mathrm{~Hz}, 18 \mathrm{~A}-\mathrm{H}), 5.12(\mathrm{dd}, 1 \mathrm{H}, J=9.8,9.8 \mathrm{~Hz}, 4 \mathrm{H}), 5.05(\mathrm{dd}, 1 \mathrm{H}, J=9.5,8.1 \mathrm{~Hz}$, $\left.2^{\prime}-\mathrm{H}\right), 5.01(\mathrm{dd}, 1 \mathrm{H}, J=11.2,3.0 \mathrm{~Hz}, 3-\mathrm{H}), 4.97\left(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 4.71(\mathrm{dd}, 1 \mathrm{H}, J=16.6,1.2 \mathrm{~Hz}, 5-\mathrm{H}), 4.32$ $\left(\mathrm{dd}, 1 \mathrm{H}, J=12.5,4.7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 4.16\left(\mathrm{dd}, 1 \mathrm{H}, J=12.5,2.2 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 3.78\left(\mathrm{ddd}, 1 \mathrm{H}, J=10.0,4.7,2.2 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right)$, $3.08(\mathrm{~m}, 1 \mathrm{H}, 15-\mathrm{H}), 2.82(\mathrm{ddd}, 1 \mathrm{H}, J=9.6,5.6,1.8 \mathrm{~Hz}, 20-\mathrm{H}), 2.66(\mathrm{ddd}, 1 \mathrm{H}, J=12.9,3.6,3.6 \mathrm{~Hz}, 14 \beta-\mathrm{H}), 2.11$, $2.04,2.02$ and 2.01 (each $\mathrm{s}, 3 \mathrm{H}, 3 \times \mathrm{OAc}$ ), $1.55(\mathrm{ddd}, 1 \mathrm{H}, J=12.9,12.3,12.3 \mathrm{~Hz}, 14 \alpha-\mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(125 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) $\delta: 161.8(\mathrm{C}-2), 61.5(\mathrm{C}-3), 48.5(\mathrm{C}-5), 128.0(\mathrm{C}-6), 130.0(\mathrm{C}-7), 127.7(\mathrm{C}-8), 127.9(\mathrm{C}-9), 126.7(\mathrm{C}-10)$, 129.5 (C-11), 129.0 (C-12), 148.2 (C-13), 30.0 (C-14), 28.3 (C-15), 108.4 (C-16), $146.8(\mathrm{C}-17), 120.8(\mathrm{C}-18), 131.5$ (C-19), $42.9(\mathrm{C}-20), 96.4(\mathrm{C}-21), 162.8(\mathrm{C}-22), 96.1\left(\mathrm{C}-1^{\prime}\right), 70.6\left(\mathrm{C}-2^{\prime}\right), 72.4\left(\mathrm{C}-3^{\prime}\right), 68.2\left(\mathrm{C}-4^{\prime}\right), 72.3\left(\mathrm{C}-5^{\prime}\right), 61.8(\mathrm{C}-$ 6'), 20.74 and 20.65 (each CO-Me), 20.57 ( $2 \times \mathrm{CO}-\mathrm{Me}$ ), 170.6 and 170.1 (each CO-Me), 169.4 ( $2 \times \mathrm{CO}-\mathrm{Me}$ ); CD (c $\left.=0.21 \mathrm{mmol} / \mathrm{l}, \mathrm{MeOH}, 21^{\circ} \mathrm{C}\right) \Delta \varepsilon(\lambda \mathrm{nm}): 0(330),+1.46(320),+0.29(317),+0.88(313),+0.29(310),+1.17$ (307), $+0.58(303),+6.56(276), 0(266),-16.62(247),-27.12(238), 0(225),-2.33(218), 0(213),+5.83(208)$.
11. $3(S)$-Deoxypumiloside tetraacetate (8) : HR-MS (NBA) $m / z: 665.2347$ (Calcd for $\mathrm{C}_{34} \mathrm{H}_{37} \mathrm{~N}_{2} \mathrm{O}_{12}: 665.2347$ ); FABMS (NBA) $m / z(\%): 665\left(\mathrm{MH}^{+}, 100\right), 154(85) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 8.08(\mathrm{~d}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}, 12-\mathrm{H}), 8.07$ $(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.82(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 9-\mathrm{H}), 7.71(\mathrm{ddd}, 1 \mathrm{H}, J=8.3,6.9,1.4 \mathrm{~Hz}, 11-\mathrm{H}), 7.56(\mathrm{ddd}, 1 \mathrm{H}, J=8.0,6.9,1.0$ $\mathrm{Hz}, 10-\mathrm{H}), 7.16(\mathrm{~d}, 1 \mathrm{H}, J=2.7 \mathrm{~Hz}, 17-\mathrm{H}), 5.80(\mathrm{ddd}, 1 \mathrm{H}, J=17.0,9.9,9.9 \mathrm{~Hz}, 19-\mathrm{H}), 5.51(\mathrm{dd}, 1 \mathrm{H}, J=17.2,1.9$ $\mathrm{Hz}, 18 \mathrm{~B}-\mathrm{H}), 5.05-4.98(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{H}), 5.29(\mathrm{~d}, 1 \mathrm{H}, J=1.9 \mathrm{~Hz}, 21-\mathrm{H}), 5.27(\mathrm{dd}, 1 \mathrm{H}, J=9.5,9.5 \mathrm{~Hz}, 3 \mathrm{H}-\mathrm{H}), 5.40(\mathrm{dd}$, $1 \mathrm{H}, J=10.2,1.6 \mathrm{~Hz}, 18 \mathrm{~A}-\mathrm{H}), 5.10\left(\mathrm{dd}, 1 \mathrm{H}, J=9.8,9.8 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.03\left(\mathrm{dd}, 1 \mathrm{H}, J=9.2,8.1 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 4.98(\mathrm{~d}, 1 \mathrm{H}, J$ $\left.=8.1 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 4.78(\mathrm{dd}, 1 \mathrm{H}, J=16.5,1.1 \mathrm{~Hz}, 5-\mathrm{H}), 4.73(\mathrm{dd}, 1 \mathrm{H}, J=7.9,4.1 \mathrm{~Hz}, 3-\mathrm{H}), 4.30(\mathrm{dd}, 1 \mathrm{H}, J=12.5,2.2$ $\left.\mathrm{Hz}, 6^{\prime}-\mathrm{H}\right), 4.16\left(\mathrm{dd}, 1 \mathrm{H}, J=12.5,4.4 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 3.77\left(\mathrm{ddd}, 1 \mathrm{H}, J=10.0,4.7,2.3 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 3.15(\mathrm{~m}, 1 \mathrm{H}, 15-\mathrm{H}), 2.72$ (ddd, $1 \mathrm{H}, J=9.5,5.3,1.4 \mathrm{~Hz}, 20-\mathrm{H}), 2.62(\mathrm{dd}, 1 \mathrm{H}, J=13.4,5.0 \mathrm{~Hz}, 14 \alpha-\mathrm{H}), 2.10(\mathrm{~m}, 1 \mathrm{H}, 14 \beta-\mathrm{H}), 2.10(\mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{x}$ OAc ), 2.04 and 2.01 (each $\mathrm{s}, 3 \mathrm{H}, 2 \times \mathrm{OAc}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 161.9(\mathrm{C}-2), 60.2(\mathrm{C}-3), 47.9(\mathrm{C}-5)$, 128.0 (C-6), 130.4 (C-7), 127.6 (C-8), 127.9 (C-9), 126.7 (C-10), 129.5 (C-11), 129.1 (C-12), 148.2 (C-13), 28.8 (C14), $23.8(\mathrm{C}-15), 110.4(\mathrm{C}-16), 145.1(\mathrm{C}-17), 121.6(\mathrm{C}-18), 131.4(\mathrm{C}-19), 43.8(\mathrm{C}-20), 96.3(\mathrm{C}-21), 165.2(\mathrm{C}-22)$, 96.1 ( $\mathrm{C}-\mathrm{I}^{\prime}$ ), 70.4 (C-2'), 72.4 (C-3'), 68.1 ( $\mathrm{C}-4^{\prime}$ ), 72.2 (C-5'), 61.7 (C-6'), 20.8 and 20.7 (each CO-Me), 20.6 ( $2 \times \mathrm{CO}-$ $M e), 170.6,170.1,169.9$ and 169.4 (each $\mathrm{CO}-\mathrm{Me}) ; \mathrm{CD}\left(\mathrm{c}=0.15 \mathrm{mmol} / \mathrm{l}, \mathrm{MeOH}, 21^{\circ} \mathrm{C}\right) \Delta \varepsilon(\lambda \mathrm{nm}): 0(325),-1.01$ (320), $-0.40(317),-0.81(313),-0.40(310),-0.60(307),-0.20(300),-25.15(238), 0(227),+3.62(218),+10.87$ (206).
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